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(E)-1-[4-[Bis(4-methoxyphenyl)methyl]-piperazin-1-yl]-3-(4-fluorophenyl)prop-2-en-1-one

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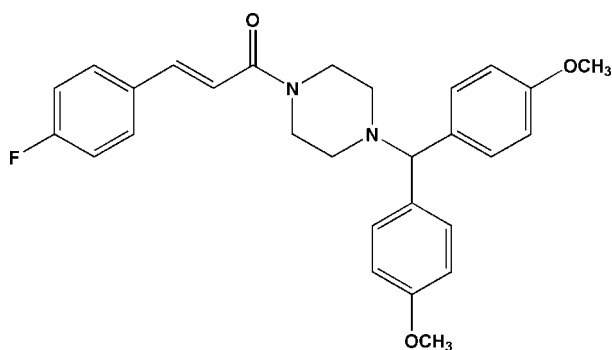
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Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; R factor = 0.061; wR factor = 0.170; data-to-parameter ratio = 14.5.

In the title compound, $\text{C}_{28}\text{H}_{29}\text{FN}_2\text{O}_3$, the conformation about the ethene bond is *E*. The piperazine ring adopts a chair conformation. In the crystal, molecules are linked by intermolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds.

Related literature

For properties of cinnamic acid derivatives, see: Shi *et al.* (2005); Point *et al.* (1998). For synthetic procedures, see: Wu *et al.* (2008). For a related structure, see: Mouillé *et al.* (1975).



Experimental

Crystal data

 $\text{C}_{28}\text{H}_{29}\text{FN}_2\text{O}_3$ $M_r = 460.53$ Monoclinic, $P2_1/c$ $a = 10.235$ (2) Å $b = 7.8420$ (16) Å $c = 30.385$ (6) Å $\beta = 96.65$ (3)° $V = 2422.4$ (8) Å³ $Z = 4$ Mo $K\alpha$ radiation $\mu = 0.09$ mm⁻¹ $T = 293$ K $0.20 \times 0.10 \times 0.10$ mm

Data collection

Enraf–Nonius CAD-4

diffractometer

Absorption correction: ψ scan(North *et al.*, 1968) $T_{\min} = 0.983$, $T_{\max} = 0.991$

4730 measured reflections

4463 independent reflections

2366 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.031$

3 standard reflections every 200 reflections

intensity decay: 1%

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.061$ $wR(F^2) = 0.170$ $S = 1.01$

4463 reflections

307 parameters

H-atom parameters constrained

 $\Delta\rho_{\max} = 0.13$ e Å⁻³ $\Delta\rho_{\min} = -0.18$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|--|-------|-------------|-------------|---------------|
| $\text{C5}-\text{H5A}\cdots\text{O1}^i$ | 0.93 | 2.28 | 3.131 (4) | 152 |
| $\text{C17}-\text{H17A}\cdots\text{O3}^{ii}$ | 0.93 | 2.60 | 3.499 (4) | 163 |

Symmetry codes: (i) $-x + 1, -y - 1, -z$; (ii) $x - 1, y, z$.

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1994); cell refinement: *CAD-4 EXPRESS*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: PV2381).

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supplementary materials

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(*E*)-1-{4-[Bis(4-methoxyphenyl)methyl]piperazin-1-yl}-3-(4-fluorophenyl)prop-2-en-1-one

Y.-B. Teng, Z.-H. Dai and B. Wu

Comment

Cinnamic acid derivatives have been reported to possess many useful properties, including alpha-glucosidase inhibition, acyl-CoA inhibition, LDL-oxidation inhibition, tyrosinase inhibition, antioxidant, antimicrobial, neuroprotective activities (Shi *et al.*, 2005; Point *et al.*, 1998). We report here the synthesis and crystal structure of a novel cinnamic acid derivative.

In the title molecule (Fig. 1), the conformation about the ethene bond C7=C8 is *E*. The piperazine ring adopts a chair conformation. There are intramolecular and intermolecular C—H···O hydrogen bonds in the title compound (Fig. 2) which consolidate the crystal structure. The bond lengths and angles in the title compound agree well with the corresponding bond lengths and angles in a closely related compound, *trans*-cinnamyl-1-diphenylmethyl-4-piperazine (Mouillé *et al.*, 1975).

Experimental

The synthesis follows the method of Wu *et al.* (2008). A mixture of (*E*)-3-(4-fluoro phenyl)acrylic acid (1.66 g; 10 mmol), dimethyl sulfoxide (4 ml) and dichloromethane (60 ml) was stirred for 6 h at room temperature. The solvent was removed under reduced pressure. The residue was dissolved in acetone (60 ml) and reacted with 1-(bis(4-methoxyphenyl)methyl) piperazine (4.69 g; 15 mmol) in the presence of triethylamine (12 ml) for 5 h at room temperature. The resultant mixture was cooled. The solid thus obtained was filtered and recrystallized from ethanol to afford the title compound. Pale-yellow single crystals of the title compound suitable for *X*-ray diffraction studies were grown from a mixture of CHCl₃ and hexane (1:1) by slow evaporation at room temperature.

Refinement

All H atoms were placed geometrically at distances C—H = 0.93, 0.96, 0.97 and 0.98 Å for aryl, methyl, methylene and methyne type H-atoms, respectively, and included in the refinement in riding motion approximation with $U_{\text{iso}}(\text{H}) = 1.2$ or $1.5U_{\text{eq}}$ of the carrier atom.

Figures

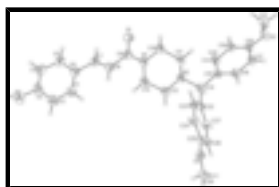


Fig. 1. Molecular structure of the title compound, showing the atom labeling scheme and 70% probability displacement ellipsoids.

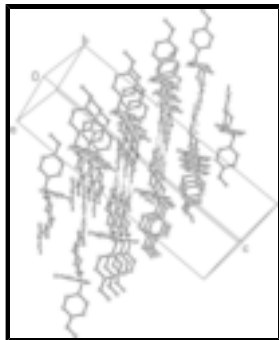


Fig. 2. Packing diagram of the title compound showing hydrogen bonds as dashed lines.

(E)-1-[4-[Bis(4-methoxyphenyl)methyl]piperazin-1-yl]-3-(4-fluorophenyl)prop-2-en-1-one

Crystal data

$C_{28}H_{29}FN_2O_3$

$M_r = 460.53$

Monoclinic, $P2_1/c$

Hall symbol: -P 2ybc

$a = 10.235 (2) \text{ \AA}$

$b = 7.8420 (16) \text{ \AA}$

$c = 30.385 (6) \text{ \AA}$

$\beta = 96.65 (3)^\circ$

$V = 2422.4 (8) \text{ \AA}^3$

$Z = 4$

$F(000) = 976$

$D_x = 1.263 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 25 reflections

$\theta = 10\text{--}13^\circ$

$\mu = 0.09 \text{ mm}^{-1}$

$T = 293 \text{ K}$

Block, pale-yellow

$0.20 \times 0.10 \times 0.10 \text{ mm}$

Data collection

Enraf-Nonius CAD-4
diffractometer

Radiation source: fine-focus sealed tube
graphite

ω and 2θ scans

Absorption correction: multi-scan
 ψ scan

$T_{\min} = 0.983$, $T_{\max} = 0.991$

4730 measured reflections

4463 independent reflections

2366 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.031$

$\theta_{\max} = 25.4^\circ$, $\theta_{\min} = 1.4^\circ$

$h = 0 \rightarrow 12$

$k = 0 \rightarrow 9$

$l = -36 \rightarrow 36$

3 standard reflections every 200 reflections

intensity decay: 1%

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.061$

$wR(F^2) = 0.170$

$S = 1.01$

Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.080P)^2]$

4463 reflections
307 parameters
0 restraints

where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.13 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.18 \text{ e } \text{\AA}^{-3}$

Special details

Experimental. (North *et al.*, 1968)

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | <i>x</i> | <i>y</i> | <i>z</i> | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|------|--------------|-------------|--------------|----------------------------------|
| F | 1.1036 (2) | -0.8047 (3) | 0.03071 (8) | 0.1138 (8) |
| N1 | 0.5564 (2) | 0.0310 (3) | 0.08123 (8) | 0.0606 (7) |
| O1 | 0.4552 (2) | -0.1690 (3) | 0.03651 (7) | 0.0783 (7) |
| C1 | 0.9101 (3) | -0.4669 (4) | 0.06924 (11) | 0.0723 (10) |
| H1A | 0.9196 | -0.3770 | 0.0893 | 0.087* |
| O2 | 0.06003 (19) | 0.8563 (3) | 0.14269 (7) | 0.0658 (6) |
| N2 | 0.5216 (2) | 0.3082 (3) | 0.13971 (8) | 0.0540 (6) |
| C2 | 1.0149 (4) | -0.5726 (5) | 0.06570 (13) | 0.0838 (11) |
| H2A | 1.0948 | -0.5563 | 0.0831 | 0.101* |
| O3 | 0.93784 (18) | 0.7764 (3) | 0.25379 (7) | 0.0653 (6) |
| C3 | 0.9982 (4) | -0.7018 (5) | 0.03591 (13) | 0.0779 (10) |
| C4 | 0.8809 (4) | -0.7385 (5) | 0.01176 (12) | 0.0841 (11) |
| H4A | 0.8712 | -0.8333 | -0.0067 | 0.101* |
| C5 | 0.7775 (4) | -0.6302 (5) | 0.01579 (11) | 0.0755 (10) |
| H5A | 0.6967 | -0.6512 | -0.0007 | 0.091* |
| C6 | 0.7909 (3) | -0.4905 (4) | 0.04381 (10) | 0.0592 (8) |
| C7 | 0.6790 (3) | -0.3740 (4) | 0.04508 (10) | 0.0636 (9) |
| H7A | 0.5982 | -0.4155 | 0.0324 | 0.076* |
| C8 | 0.6773 (3) | -0.2193 (4) | 0.06162 (10) | 0.0619 (9) |
| H8A | 0.7551 | -0.1731 | 0.0755 | 0.074* |
| C9 | 0.5553 (3) | -0.1162 (4) | 0.05889 (10) | 0.0574 (8) |
| C10 | 0.4434 (3) | 0.1440 (4) | 0.07401 (10) | 0.0677 (9) |
| H10A | 0.3688 | 0.0814 | 0.0595 | 0.081* |
| H10B | 0.4629 | 0.2365 | 0.0546 | 0.081* |
| C11 | 0.4086 (3) | 0.2162 (4) | 0.11695 (10) | 0.0633 (9) |
| H11A | 0.3345 | 0.2932 | 0.1112 | 0.076* |
| H11B | 0.3834 | 0.1246 | 0.1357 | 0.076* |

supplementary materials

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|------|------------|------------|--------------|-------------|
| C12 | 0.6265 (3) | 0.1835 (4) | 0.15010 (11) | 0.0662 (9) |
| H12A | 0.5962 | 0.0944 | 0.1686 | 0.079* |
| H12B | 0.7016 | 0.2387 | 0.1666 | 0.079* |
| C13 | 0.6676 (3) | 0.1057 (4) | 0.10871 (11) | 0.0658 (9) |
| H13A | 0.7075 | 0.1927 | 0.0919 | 0.079* |
| H13B | 0.7331 | 0.0182 | 0.1167 | 0.079* |
| C14 | 0.4907 (3) | 0.4005 (4) | 0.17929 (9) | 0.0539 (8) |
| H14A | 0.4654 | 0.3170 | 0.2008 | 0.065* |
| C15 | 0.3763 (3) | 0.5219 (4) | 0.16789 (10) | 0.0496 (7) |
| C16 | 0.2791 (3) | 0.5383 (4) | 0.19576 (10) | 0.0562 (8) |
| H16A | 0.2837 | 0.4723 | 0.2213 | 0.067* |
| C17 | 0.1757 (3) | 0.6503 (4) | 0.18645 (10) | 0.0588 (8) |
| H17A | 0.1113 | 0.6587 | 0.2056 | 0.071* |
| C18 | 0.1677 (3) | 0.7490 (4) | 0.14916 (10) | 0.0517 (7) |
| C19 | 0.2627 (3) | 0.7373 (4) | 0.12093 (10) | 0.0564 (8) |
| H19A | 0.2579 | 0.8045 | 0.0956 | 0.068* |
| C20 | 0.3658 (3) | 0.6239 (4) | 0.13073 (10) | 0.0576 (8) |
| H20A | 0.4302 | 0.6165 | 0.1116 | 0.069* |
| C21 | 0.0545 (3) | 0.9718 (4) | 0.10672 (11) | 0.0739 (10) |
| H21A | -0.0246 | 1.0382 | 0.1056 | 0.111* |
| H21B | 0.0548 | 0.9093 | 0.0796 | 0.111* |
| H21C | 0.1295 | 1.0461 | 0.1106 | 0.111* |
| C22 | 0.6118 (3) | 0.4964 (4) | 0.20031 (9) | 0.0487 (7) |
| C23 | 0.6794 (3) | 0.6068 (4) | 0.17551 (10) | 0.0654 (9) |
| H23A | 0.6512 | 0.6209 | 0.1455 | 0.078* |
| C24 | 0.7869 (3) | 0.6956 (4) | 0.19426 (10) | 0.0640 (9) |
| H24A | 0.8311 | 0.7679 | 0.1768 | 0.077* |
| C25 | 0.8303 (3) | 0.6793 (4) | 0.23854 (10) | 0.0516 (7) |
| C26 | 0.7663 (3) | 0.5699 (4) | 0.26369 (10) | 0.0594 (8) |
| H26A | 0.7956 | 0.5555 | 0.2936 | 0.071* |
| C27 | 0.6569 (3) | 0.4799 (4) | 0.24439 (10) | 0.0583 (8) |
| H27A | 0.6134 | 0.4066 | 0.2619 | 0.070* |
| C28 | 0.9994 (3) | 0.7437 (5) | 0.29707 (11) | 0.0853 (11) |
| H28A | 1.0722 | 0.8203 | 0.3037 | 0.128* |
| H28B | 1.0305 | 0.6282 | 0.2989 | 0.128* |
| H28C | 0.9372 | 0.7606 | 0.3180 | 0.128* |

Atomic displacement parameters (\AA^2)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|----|-------------|-------------|-------------|--------------|--------------|--------------|
| F | 0.1173 (18) | 0.1002 (17) | 0.1298 (19) | 0.0444 (15) | 0.0391 (15) | 0.0195 (15) |
| N1 | 0.0469 (15) | 0.0636 (17) | 0.0693 (17) | 0.0060 (14) | -0.0021 (13) | -0.0110 (15) |
| O1 | 0.0608 (14) | 0.0883 (17) | 0.0834 (16) | -0.0077 (13) | -0.0015 (12) | -0.0212 (13) |
| C1 | 0.075 (2) | 0.053 (2) | 0.084 (2) | 0.0013 (19) | -0.0119 (19) | -0.0117 (18) |
| O2 | 0.0542 (12) | 0.0670 (14) | 0.0768 (15) | 0.0124 (12) | 0.0106 (10) | 0.0043 (12) |
| N2 | 0.0361 (13) | 0.0562 (15) | 0.0682 (16) | 0.0014 (12) | -0.0003 (11) | -0.0075 (13) |
| C2 | 0.075 (2) | 0.059 (2) | 0.113 (3) | 0.006 (2) | -0.010 (2) | -0.003 (2) |
| O3 | 0.0502 (12) | 0.0750 (15) | 0.0689 (14) | -0.0098 (12) | -0.0008 (10) | -0.0058 (12) |

| | | | | | | |
|-----|-------------|-------------|-------------|--------------|--------------|--------------|
| C3 | 0.087 (3) | 0.067 (3) | 0.083 (3) | 0.019 (2) | 0.026 (2) | 0.017 (2) |
| C4 | 0.109 (3) | 0.079 (3) | 0.066 (2) | 0.015 (3) | 0.017 (2) | -0.018 (2) |
| C5 | 0.081 (2) | 0.080 (3) | 0.065 (2) | -0.004 (2) | 0.0024 (18) | -0.019 (2) |
| C6 | 0.067 (2) | 0.0534 (19) | 0.0571 (18) | -0.0045 (17) | 0.0063 (16) | -0.0043 (16) |
| C7 | 0.059 (2) | 0.069 (2) | 0.062 (2) | -0.0046 (18) | 0.0049 (16) | -0.0068 (18) |
| C8 | 0.0550 (19) | 0.060 (2) | 0.070 (2) | -0.0064 (17) | 0.0063 (15) | -0.0123 (18) |
| C9 | 0.0530 (19) | 0.063 (2) | 0.0568 (19) | -0.0047 (17) | 0.0098 (15) | -0.0026 (17) |
| C10 | 0.0502 (18) | 0.075 (2) | 0.074 (2) | 0.0085 (18) | -0.0097 (16) | -0.0102 (18) |
| C11 | 0.0391 (16) | 0.067 (2) | 0.081 (2) | 0.0023 (16) | -0.0036 (15) | -0.0091 (18) |
| C12 | 0.0475 (18) | 0.063 (2) | 0.085 (2) | 0.0043 (16) | -0.0082 (16) | -0.0163 (18) |
| C13 | 0.0429 (17) | 0.064 (2) | 0.089 (2) | 0.0022 (16) | 0.0026 (16) | -0.0136 (19) |
| C14 | 0.0489 (17) | 0.0533 (18) | 0.0598 (19) | -0.0024 (15) | 0.0077 (14) | 0.0045 (16) |
| C15 | 0.0384 (15) | 0.0494 (17) | 0.0606 (18) | -0.0038 (14) | 0.0045 (13) | -0.0015 (15) |
| C16 | 0.0540 (18) | 0.0579 (19) | 0.0587 (18) | -0.0039 (16) | 0.0150 (15) | 0.0081 (16) |
| C17 | 0.0470 (17) | 0.065 (2) | 0.066 (2) | 0.0046 (16) | 0.0166 (15) | 0.0020 (17) |
| C18 | 0.0414 (16) | 0.0525 (18) | 0.0610 (19) | -0.0016 (15) | 0.0059 (14) | -0.0063 (16) |
| C19 | 0.0553 (18) | 0.0559 (19) | 0.0588 (19) | 0.0031 (16) | 0.0096 (15) | 0.0058 (16) |
| C20 | 0.0430 (16) | 0.067 (2) | 0.066 (2) | 0.0037 (16) | 0.0170 (14) | 0.0057 (17) |
| C21 | 0.059 (2) | 0.074 (2) | 0.088 (3) | 0.0111 (18) | 0.0009 (18) | 0.006 (2) |
| C22 | 0.0414 (15) | 0.0493 (17) | 0.0557 (18) | 0.0038 (14) | 0.0063 (13) | -0.0022 (15) |
| C23 | 0.062 (2) | 0.085 (2) | 0.0485 (18) | -0.0180 (19) | 0.0023 (15) | 0.0081 (17) |
| C24 | 0.0528 (18) | 0.079 (2) | 0.061 (2) | -0.0176 (18) | 0.0077 (15) | 0.0101 (18) |
| C25 | 0.0406 (16) | 0.0512 (18) | 0.063 (2) | 0.0044 (15) | 0.0061 (15) | -0.0050 (16) |
| C26 | 0.0550 (18) | 0.073 (2) | 0.0476 (17) | 0.0007 (18) | -0.0041 (15) | 0.0037 (16) |
| C27 | 0.0533 (18) | 0.062 (2) | 0.0598 (19) | -0.0049 (17) | 0.0067 (15) | 0.0113 (17) |
| C28 | 0.063 (2) | 0.100 (3) | 0.086 (3) | -0.003 (2) | -0.0217 (19) | 0.003 (2) |

Geometric parameters (Å, °)

| | | | |
|--------|-----------|----------|-----------|
| F—C3 | 1.370 (4) | C12—H12A | 0.9700 |
| N1—C9 | 1.338 (4) | C12—H12B | 0.9700 |
| N1—C10 | 1.453 (4) | C13—H13A | 0.9700 |
| N1—C13 | 1.455 (4) | C13—H13B | 0.9700 |
| O1—C9 | 1.235 (3) | C14—C15 | 1.517 (4) |
| C1—C2 | 1.370 (4) | C14—C22 | 1.525 (4) |
| C1—C6 | 1.379 (4) | C14—H14A | 0.9800 |
| C1—H1A | 0.9300 | C15—C20 | 1.378 (4) |
| O2—C18 | 1.382 (3) | C15—C16 | 1.385 (4) |
| O2—C21 | 1.416 (3) | C16—C17 | 1.380 (4) |
| N2—C12 | 1.459 (3) | C16—H16A | 0.9300 |
| N2—C11 | 1.466 (3) | C17—C18 | 1.367 (4) |
| N2—C14 | 1.469 (3) | C17—H17A | 0.9300 |
| C2—C3 | 1.357 (5) | C18—C19 | 1.372 (4) |
| C2—H2A | 0.9300 | C19—C20 | 1.386 (4) |
| O3—C25 | 1.374 (3) | C19—H19A | 0.9300 |
| O3—C28 | 1.414 (3) | C20—H20A | 0.9300 |
| C3—C4 | 1.364 (5) | C21—H21A | 0.9600 |
| C4—C5 | 1.373 (5) | C21—H21B | 0.9600 |
| C4—H4A | 0.9300 | C21—H21C | 0.9600 |

supplementary materials

| | | | |
|------------|-----------|---------------|-----------|
| C5—C6 | 1.385 (4) | C22—C27 | 1.372 (4) |
| C5—H5A | 0.9300 | C22—C23 | 1.384 (4) |
| C6—C7 | 1.469 (4) | C23—C24 | 1.370 (4) |
| C7—C8 | 1.314 (4) | C23—H23A | 0.9300 |
| C7—H7A | 0.9300 | C24—C25 | 1.373 (4) |
| C8—C9 | 1.482 (4) | C24—H24A | 0.9300 |
| C8—H8A | 0.9300 | C25—C26 | 1.365 (4) |
| C10—C11 | 1.503 (4) | C26—C27 | 1.394 (4) |
| C10—H10A | 0.9700 | C26—H26A | 0.9300 |
| C10—H10B | 0.9700 | C27—H27A | 0.9300 |
| C11—H11A | 0.9700 | C28—H28A | 0.9600 |
| C11—H11B | 0.9700 | C28—H28B | 0.9600 |
| C12—C13 | 1.501 (4) | C28—H28C | 0.9600 |
| C9—N1—C10 | 119.3 (2) | C12—C13—H13B | 109.3 |
| C9—N1—C13 | 126.8 (3) | H13A—C13—H13B | 108.0 |
| C10—N1—C13 | 113.4 (2) | N2—C14—C15 | 110.8 (2) |
| C2—C1—C6 | 121.7 (3) | N2—C14—C22 | 110.1 (2) |
| C2—C1—H1A | 119.2 | C15—C14—C22 | 110.8 (2) |
| C6—C1—H1A | 119.2 | N2—C14—H14A | 108.3 |
| C18—O2—C21 | 117.3 (2) | C15—C14—H14A | 108.3 |
| C12—N2—C11 | 107.0 (2) | C22—C14—H14A | 108.3 |
| C12—N2—C14 | 112.1 (2) | C20—C15—C16 | 117.0 (3) |
| C11—N2—C14 | 113.3 (2) | C20—C15—C14 | 122.4 (3) |
| C3—C2—C1 | 117.7 (3) | C16—C15—C14 | 120.6 (3) |
| C3—C2—H2A | 121.1 | C17—C16—C15 | 121.5 (3) |
| C1—C2—H2A | 121.1 | C17—C16—H16A | 119.3 |
| C25—O3—C28 | 117.8 (3) | C15—C16—H16A | 119.3 |
| C2—C3—C4 | 123.5 (4) | C18—C17—C16 | 120.1 (3) |
| C2—C3—F | 118.5 (4) | C18—C17—H17A | 120.0 |
| C4—C3—F | 118.0 (4) | C16—C17—H17A | 120.0 |
| C3—C4—C5 | 117.5 (3) | C17—C18—C19 | 120.1 (3) |
| C3—C4—H4A | 121.3 | C17—C18—O2 | 115.6 (3) |
| C5—C4—H4A | 121.3 | C19—C18—O2 | 124.3 (3) |
| C4—C5—C6 | 121.5 (3) | C18—C19—C20 | 119.1 (3) |
| C4—C5—H5A | 119.3 | C18—C19—H19A | 120.5 |
| C6—C5—H5A | 119.3 | C20—C19—H19A | 120.5 |
| C1—C6—C5 | 118.0 (3) | C15—C20—C19 | 122.3 (3) |
| C1—C6—C7 | 122.9 (3) | C15—C20—H20A | 118.9 |
| C5—C6—C7 | 119.1 (3) | C19—C20—H20A | 118.9 |
| C8—C7—C6 | 129.0 (3) | O2—C21—H21A | 109.5 |
| C8—C7—H7A | 115.5 | O2—C21—H21B | 109.5 |
| C6—C7—H7A | 115.5 | H21A—C21—H21B | 109.5 |
| C7—C8—C9 | 122.1 (3) | O2—C21—H21C | 109.5 |
| C7—C8—H8A | 119.0 | H21A—C21—H21C | 109.5 |
| C9—C8—H8A | 119.0 | H21B—C21—H21C | 109.5 |
| O1—C9—N1 | 121.8 (3) | C27—C22—C23 | 117.2 (3) |
| O1—C9—C8 | 119.2 (3) | C27—C22—C14 | 121.9 (3) |
| N1—C9—C8 | 119.0 (3) | C23—C22—C14 | 120.9 (3) |
| N1—C10—C11 | 111.3 (2) | C24—C23—C22 | 121.3 (3) |

| | | | |
|----------------|------------|-----------------|------------|
| N1—C10—H10A | 109.4 | C24—C23—H23A | 119.3 |
| C11—C10—H10A | 109.4 | C22—C23—H23A | 119.3 |
| N1—C10—H10B | 109.4 | C23—C24—C25 | 120.9 (3) |
| C11—C10—H10B | 109.4 | C23—C24—H24A | 119.6 |
| H10A—C10—H10B | 108.0 | C25—C24—H24A | 119.6 |
| N2—C11—C10 | 110.0 (2) | C26—C25—C24 | 119.0 (3) |
| N2—C11—H11A | 109.7 | C26—C25—O3 | 125.2 (3) |
| C10—C11—H11A | 109.7 | C24—C25—O3 | 115.8 (3) |
| N2—C11—H11B | 109.7 | C25—C26—C27 | 119.8 (3) |
| C10—C11—H11B | 109.7 | C25—C26—H26A | 120.1 |
| H11A—C11—H11B | 108.2 | C27—C26—H26A | 120.1 |
| N2—C12—C13 | 111.2 (3) | C22—C27—C26 | 121.8 (3) |
| N2—C12—H12A | 109.4 | C22—C27—H27A | 119.1 |
| C13—C12—H12A | 109.4 | C26—C27—H27A | 119.1 |
| N2—C12—H12B | 109.4 | O3—C28—H28A | 109.5 |
| C13—C12—H12B | 109.4 | O3—C28—H28B | 109.5 |
| H12A—C12—H12B | 108.0 | H28A—C28—H28B | 109.5 |
| N1—C13—C12 | 111.6 (2) | O3—C28—H28C | 109.5 |
| N1—C13—H13A | 109.3 | H28A—C28—H28C | 109.5 |
| C12—C13—H13A | 109.3 | H28B—C28—H28C | 109.5 |
| N1—C13—H13B | 109.3 | | |
| C6—C1—C2—C3 | 0.4 (5) | N2—C14—C15—C20 | -43.6 (4) |
| C1—C2—C3—C4 | -4.6 (6) | C22—C14—C15—C20 | 79.0 (3) |
| C1—C2—C3—F | 177.5 (3) | N2—C14—C15—C16 | 138.6 (3) |
| C2—C3—C4—C5 | 4.9 (6) | C22—C14—C15—C16 | -98.9 (3) |
| F—C3—C4—C5 | -177.2 (3) | C20—C15—C16—C17 | 0.7 (4) |
| C3—C4—C5—C6 | -1.0 (5) | C14—C15—C16—C17 | 178.7 (3) |
| C2—C1—C6—C5 | 3.1 (5) | C15—C16—C17—C18 | -0.4 (4) |
| C2—C1—C6—C7 | -176.5 (3) | C16—C17—C18—C19 | -0.1 (4) |
| C4—C5—C6—C1 | -2.8 (5) | C16—C17—C18—O2 | 179.6 (2) |
| C4—C5—C6—C7 | 176.9 (3) | C21—O2—C18—C17 | 174.3 (3) |
| C1—C6—C7—C8 | 15.0 (5) | C21—O2—C18—C19 | -5.9 (4) |
| C5—C6—C7—C8 | -164.7 (3) | C17—C18—C19—C20 | 0.2 (4) |
| C6—C7—C8—C9 | 178.1 (3) | O2—C18—C19—C20 | -179.5 (3) |
| C10—N1—C9—O1 | -9.2 (4) | C16—C15—C20—C19 | -0.6 (4) |
| C13—N1—C9—O1 | 179.7 (3) | C14—C15—C20—C19 | -178.6 (3) |
| C10—N1—C9—C8 | 171.6 (3) | C18—C19—C20—C15 | 0.2 (4) |
| C13—N1—C9—C8 | 0.6 (5) | N2—C14—C22—C27 | -128.0 (3) |
| C7—C8—C9—O1 | -7.7 (5) | C15—C14—C22—C27 | 109.0 (3) |
| C7—C8—C9—N1 | 171.5 (3) | N2—C14—C22—C23 | 53.2 (4) |
| C9—N1—C10—C11 | 137.5 (3) | C15—C14—C22—C23 | -69.8 (3) |
| C13—N1—C10—C11 | -50.3 (4) | C27—C22—C23—C24 | 0.0 (5) |
| C12—N2—C11—C10 | -63.1 (3) | C14—C22—C23—C24 | 178.9 (3) |
| C14—N2—C11—C10 | 172.9 (2) | C22—C23—C24—C25 | -0.7 (5) |
| N1—C10—C11—N2 | 57.9 (3) | C23—C24—C25—C26 | 1.4 (5) |
| C11—N2—C12—C13 | 62.0 (3) | C23—C24—C25—O3 | -179.1 (3) |
| C14—N2—C12—C13 | -173.2 (2) | C28—O3—C25—C26 | 9.6 (4) |
| C9—N1—C13—C12 | -139.8 (3) | C28—O3—C25—C24 | -169.8 (3) |
| C10—N1—C13—C12 | 48.7 (4) | C24—C25—C26—C27 | -1.4 (4) |

supplementary materials

| | | | |
|----------------|------------|-----------------|------------|
| N2—C12—C13—N1 | -55.2 (4) | O3—C25—C26—C27 | 179.2 (3) |
| C12—N2—C14—C15 | -176.8 (2) | C23—C22—C27—C26 | 0.0 (4) |
| C11—N2—C14—C15 | -55.6 (3) | C14—C22—C27—C26 | -178.9 (3) |
| C12—N2—C14—C22 | 60.3 (3) | C25—C26—C27—C22 | 0.7 (5) |
| C11—N2—C14—C22 | -178.5 (2) | | |

Hydrogen-bond geometry (\AA , $^\circ$)

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|------------------------------------|-------|-------------|-------------|---------------|
| C5—H5A \cdots O1 ⁱ | 0.93 | 2.28 | 3.131 (4) | 152 |
| C17—H17A \cdots O3 ⁱⁱ | 0.93 | 2.60 | 3.499 (4) | 163 |
| C10—H10A \cdots O1 | 0.97 | 2.30 | 2.715 (4) | 105 |
| C7—H7A \cdots O1 | 0.93 | 2.43 | 2.786 (4) | 102 |

Symmetry codes: (i) $-x+1, -y-1, -z$; (ii) $x-1, y, z$.

Fig. 1

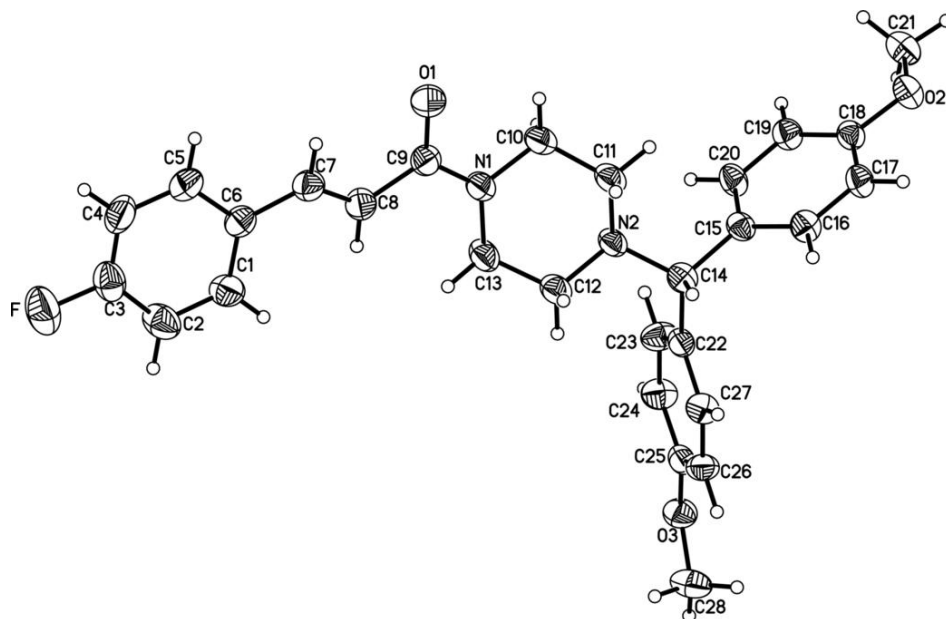


Fig. 2

